CHEMICAL STUDIES OF MARINE INVERTEBRATES. X. LOBOPHYTOLIDE, A NEW CEMBRANOLIDE DITERPENE FROM THE SOFT CORAL LOBOPHYTUM CRISTAGALLI (Coelenterata, Octocorallia, Alcyonacea) ⁽¹⁾.

B. TURSCH, J.C. BRAEKMAN[×], D. DALOZE, M. HERIN, R. KARLSSON^{××}.

Collectif de Bio-écologie Université Libre de Bruxelles Bruxelles Belgique.

(Received in UK 9 September 1974; accepted for publication 18 September 1974)

In the course of our continuing search for terpenoids from marine invertebrates, we have isolated a new cembranolide diterpene, lobophytolide (I), from the soft coral *Lobophytum cristagalli* Von Marenzeller, 1886.

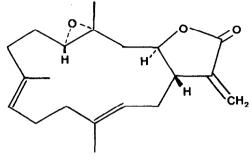
The dichloromethane extract of the sun-dried animals, collected off Serwaru, Leti Island, Indonesia ⁽²⁾, was chromatographed on silica gel. Elution with hexane/acetone (8:2) afforded crystalline lobophytolide (I ; 1% on the basis of dry starting material ; m.p. 137-8°; [a] $_{\rm D}^{\rm CHC1}$ 3 = 7° (c = 0.42) ; M = 316 ; $_{\rm 20}^{\rm H}$ ₂₆ $_{\rm 3}$). The IR ($\nu_{\rm C=0}$ at 1755 cm⁻¹ and $\nu_{\rm C=C}$ at 1670 cm⁻¹), UV ($\lambda_{\rm max}^{\rm MeOH}$ 209 nm , ε = 9080) and NMR spectra (two IH doublets at 6.22 (J = 2.5 Hz) and 5.58 ppm (J = 2 Hz) respectively) of I strongly suggest the presence of a γ -lactone conjugated to an exomethylene group ^(3,4).

This was further confirmed by NaBH₄ treatment of I (MeOH - room temperature - 3 min) which, on selective reduction of the conjugated exomethylene group ⁽⁵⁾,led to a mixture of two epimeric dihydroderivatives (II ; M = 318 ; $v_{C=0}$ at 1775 cm⁻¹ ; NMR ⁽⁶⁾ : no signals beyond 5.5 ppm but presence of two doublets (J = 7 Hz) of 1.5 H each at 1.18 and 1.25 ppm respectively).

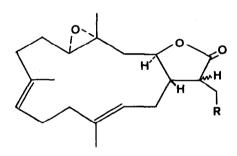
Moreover treatment of lobophytolide with a 1% solution of NaOH in methanol yielded compound III (m.p. 104-5°; M = 348; $v_{C=0}$ at 1780 cm⁻¹; NMR⁽⁶⁾: 2H doublet at 3.7 ppm and 3H singlet at 3.4 ppm (CH₂-OCH₃)) resulting from the addition of methanol to the exomethylene group⁽⁷⁾. The five membered ring nature of the conjugated lactone was demonstrated by NMR decoupling experiments. Indeed, the lH multiplet at 2.66 ppm in the spectrum of I and attributed to H₁, was found to be coupled with both the lactonic proton appearing

[×]Chercheur qualifié du Fonds National de la Recherche Scientifique.

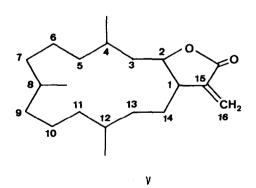
^{**} Arrhenius Laboratory, University of Stockholm, Sweden.

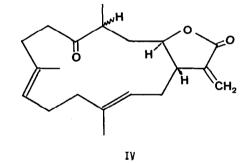


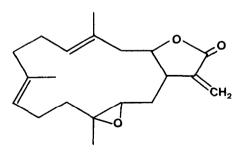
I Lobophytolide



II R = H III $R = OCH_3$







VI

as a quadruplet at 3.94 ppm ($J_{H_1H_2} = 5$ Hz) and the exomethylenic protons ($J_{H_1H_1} = 2$ Hz; $J_{H_1H_1B6cis} = 2.5$ Hz).

Since lobophytolide is devoid of either hydroxyl or further carbonyl group, the third oxygen atom of the molecule must be implied in an ether. This ether function is an epoxide since I readily reacts with p-toluenesulfonic acid to give a mixture from which the ketone (IV) could be isolated (M = 316; $v_{C=0}$ at 1765 and 1705 cm⁻¹ and $v_{C=C}$ at 1665 cm⁻¹; NMR ⁽³⁾; 3H doublet (J = 7 Hz) at 1.22 ppm (CH₃-CH-C=O)). Moreover, the presence in the NMR spectrum⁽³⁾ of I, of a 3H singlet at 1.29 ppm and a triplet of 1H (J = 6 Hz) at 2.80 ppm implies the sequence $CH_3-C_*CH-CH_2-$.

In addition to the characteristic signals previously discussed, the 270 MHz NMR spectrum of I also shows a 2H multiplet at 5.02 ppm and two 3H broad singlets at 1.66 and 1.56 ppm respectively, assignable to two CH=C-CH₃ groups. The presence of these two double bonds was further proved by the obtention of a tetrahydroderivative (M = 352; $v_{C=0}$ at 1775 cm⁻¹; NMR ⁽⁶⁾: complex 6H signal at 1.00 ppm) on hydrogenation of III over platinum oxide.

The existence of a γ -lactone conjugated to an exomethylene, two double bonds and an epoxide, accounts for 6 of the 7 unsaturations required by formula $C_{20}H_{28}O_3$ of I and implies that lobophytolide is monocarbocyclic ⁽⁸⁾. By analogy with other diterpenes isolated from Coelenterates ^(11,12) it appeared thus logical to admit that lobophytolide possesses the cembranolide skeleton (V) in which two of the three methyl groups are located on a double bond, the last one bearing the epoxide. The number of possible arrangements of these functions (initially 24) could be rapidly reduced to two - (I) and (VI) - since:

- a) microozonolysis ⁽⁹⁾ of I yields levulinaldehyde.
- b) decoupling experiments on I ⁽³⁾ show that there is no coupling between the lactonic proton at 3.94 ppm (H₂), the ether proton at 2.82 ppm (H₅) and the olefinic protons at 5.01 ppm (H₉ and H₁₃).
 c) the NMR spectrum ⁽³⁾ of the ketone IV does not show any signal beyond 3 ppm,
- c) the NMR spectrum ⁽³⁾ of the ketone IV does not show any signal beyond 3 ppm, attribuable to a methylene group located between a keto group and a double bond.

Of the two remaining structures, I was the most probable, since on irradiation at 2.66 ppm (proton H_1) one observes small modifications of the shape of the complex signal of 8H centered at 2.2 ppm and attributed to the CH₂ adjacent to sp² carbon atoms.

Independent X-ray diffraction analysis ⁽¹⁰⁾ established the structure and relative configuration of lobophytolide as I, in full agreement with the hypothesis derived from chemical and spectroscopic methods. Crystal data: lobophytolide crystallized in the orthorhombic system ; space group $P2_12_12_1$ with a = 22.29 Å, b = 13.84 Å, c = 6.00 Å.

REFERENCES

- 1- Part IX : M. KAISIN, Y.M. SHEIKH, L.J. DURHAM, C. DJERASSI, B. TURSCH, D. DALOZE, J.C. BRAEKMAN, D. LOSMAN and R. KARLSSON, Tetrahedron Letters, 2239 (1974).
- 2- Part of an extensive collection of Indonesian octocorallians effected by Mr. J. PIERRET.
- 3- 270 MHz proton spectrum in CDCl, with TMS as internal standard.

4- K. YAMAGUCHI "Spectral data of natural products" Vol 1, Ed. Elsevier (1970).

5- C, DJERASSI and W. RITTEL, J. Am. Chem. Soc., 79, 3528 (1957).

- 6- 60 MHz proton spectrum in CDCl, with TMS as internal standard.
- 7- S.M. KUPCHAN, T.J. GIACOBBE, I.S. KRULL, A.M. JHOMAS, M.A. EAKIN and D.C. FESSLER, J. Org. Chem., 35, 3539 (1970).
- 8- The presence of a tetrasubstituted double bond could be excluded since no further sp^2 carbon atom than those implicated in the functions previously described, is observed in the ¹³C NMR spectrum of I.
- 9- B.P. MOORE and W.V. BROWN, J. Chrom., 60, 157 (1971).
- 10- A full paper on the X-ray work will be published later.
- 11- L.S. CIERESZKO and T.K.B. KARNS "Biology and geology of coral reefs" Vol II, p. 183, Ed. Ac. Press (1973).
- 12- F.J. SCHMITZ, D.J. VANDERAH and L.S. CIERESZKO, Chem. Comm., 407 (1974).

Acknowledgments,

This work has been effected under NATO Special Research Grant ES-003. One of us (M.H.) is indebted to the "Institut pour l'encouragement à la Recherche Scientifique dans l'Industrie et l'Agriculture" for financial help. We thank the Indonesian Institute of Science (L.I.P.I.) for its kind cooperation. We are grateful to Professor P. KIERKEGAARD for the use of his diffractometer. We also thank Dr. D. ZIMMERMANN for the ¹³C NMR spectrum, Mr. J. PIERREI for the collection of the alcyonarians and Dr. J. VERSEVELDI for their identification.